CONFORMATION AND ORIENTATION OF GRAMICIDIN A IN ORIENTED PHOSPHOLIPID BILAYERS MEASURED BY SOLID STATE CARBON-13 NMR

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ABSTRACT Three analogues of the helical ionophore gramicidin A have been synthesized with ¹³C-labeled carbonyls (13C=O) incorporated at either Gly², Ala³, or Val⁷. A fourth compound incorporated ¹³C at both the carbonyl and α -carbon of Gly² within the same molecule. These labels were studied using solid-state, proton-enhanced, ¹³C nuclear magnetic resonance (NMR) in hydrated dispersions of dimyristoylphosphatidylcholine (DMPC)-gramicidin A. The dispersions were aligned on glass coverslips whose orientation to the magnetic field could be varied through 180°. The orientation dependence of the NMR spectrum was used to obtain an accurate measurement of the ¹³C chemical shift anisotropy (CSA), and in the case of the fourth compound, the ¹³C—¹³C dipolar coupling constant. From the measured CSA and estimates of the orientation of the ¹³C shielding tensor, we are able to determine the direction of the ¹³C—O bonds and to compare these with the predictions of the various reported models for the configuration of gramicidin A in phospholipid bilayers. Our results are consistent with the left-handed ππ^{6.3}LD single-stranded helix (Urry, D. W., J. T. Walker, and T. L. Trapane. 1982. J. Membr. Biol. 69:225-231). The right-handed $\pi\pi^{63}_{LD}$ single-stranded helix observed for gramicidin A in sodium dodecyl sulfate micelles (Arseniev, A. S., I. L. Barsukov, V. F. Bystrov, A. L. Loize, and Yu A. Ovchinnikov. 1985. FEBS (Fed. Eur. Biochem. Soc.) Lett. 186:168-174) yields a poorer fit to the data. However, the width of the carbonyl resonances suggests a distribution of molecular geometries possibly resulting from a spread in the helix pitch and handedness. Double-stranded helices and β sheet structures are excluded. In dispersions in which the lipid is in the L_a phase, the gramicidin A undergoes rapid reorientation about an axis which is centered on the normal to the plane of the coverslips. When the supporting lipid is in the L_{ff} phase the helices are rigid on the timescale of ¹³C-NMR. The configuration of gramicidin A is unaltered by L_a-L_g phase transition of the bilayer lipid.

INTRODUCTION

Gramicidin A is a low molecular weight (1,882 mol wt) ionophore that selectively facilitates the transport of small monovalent cations across lipid bilayers. It is one of a family of linear polypeptides produced during active growth of the microorganism Bacillus brevis (Katz and Demain, 1977). Many research themes have emerged from the study of gramicidin A. Because of the volume of publications reporting on the properties and structure of gramicidin A only a representative list is given here. These include the energetics of ion transport across membranes (Hladky and Haydon, 1972; Monoi, 1983; Eisenman and Sandblom, 1984; Jordan, 1984; Mackay et al., 1984; Urry et al., 1984; Etchebest et al., 1985; Kim et al., 1985; Prasad et al., 1986); the effect of gramicidin A on the order, dynamics, and phase stability of the supporting lipid (Chapman et al., 1977; Rice and Oldfield, 1979; Haigh et

al., 1979; London and Feigenson, 1981; Rajan et al., 1981; Van Echteld et al., 1981; Cavatorta et al., 1982; Cornell and Keniry, 1983; Killian and de Kruijff, 1985 a, b; Tanaka and Freed, 1985; Killian et al., 1986); and the effect of environment on the secondary and tertiary structure of gramicidin A (Urry, 1971; Veatch and Blout, 1974; Bamberg et al., 1977; Koeppe et al., 1978; Weinstein et al., 1979, 1980, 1985; Wallace et al., 1981; Urry et al., 1982, 1983 a, b; Nabedryk et al., 1982; Sychev and Ivanov, 1982, 1984; Venkatachalam and Urry, 1983; Wallace, 1984, 1986; Naik and Krimm, 1984; Hawkes et al., 1984; Koeppe and Schoenborn, 1984; Arseniev et al., 1985 a, b). The combined results of this work have made gramicidin A one of the best understood examples of a transmembrane ion channel. Despite the extensive literature on gramicidin A its conformation in phospholipid bilayers remains uncertain.

In the present study we use cross-polarization ¹³C-nuclear magnetic resonance (NMR) to derive the orientation of the ¹³C—O bonds of the Gly², Ala³, and Val⁷

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residues of gramicidin A dispersed in aligned multilayers of dimyristoyl- and dimyristylphosphatidylcholine. From these bond directions and the molecular symmetry, it is possible to discount some of the structures that have been proposed for gramicidin A in phospholipid bilayers.

MATERIALS AND METHODS

Chemicals

Butyloxycarbonyl (BOC)-amino acids and BOC-trp-O-resin were purchased from Peninsular Laboratories, San Carlos, CA. 1-13C-L-Alanine (90% enrichment) and 1-13C-L-valine (99% enrichment) were obtained from the Commissariat a L'Energy Atomique (Saclay, France). 1,2-¹³C-C₂-Glycine (90% enrichment) was obtained from Merck, Sharpe and Dohme Isotopes (Montreal, Canada), and 1-13C glycine (90% enrichment) from KOR Isotopes (Cambridge, MA). Amino protection of the ¹³C amino acids was effected by addition of the butyloxycarbonyl group using di-t-butyl dicarbonate (Sigma Chemical Co., St Louis, MO) (Prasad et al., 1982) or BOC-ON (2-[tert-butoxycarbonyloxyamino]-2-phenylacetonitrile; Ega Chemie, Albuch, FDR), following the method of Itoh et al. (1975). Dimyristoylphosphatidylcholine (DMPC) was purchased from Calbiochem-Behring Corp., La Jolla, CA, and the chain perdeuterated analogue, dimyristolyphosphatidylcholine (dsa-DMPC) was purchased from Sedary Research Laboratories Inc., London, Ontario, Canada.

Synthesis of ¹³C-labeled Peptides

Gramicidin A was synthesized using the solid-phase method, with minor variations from the procedure described by Prasad et al. (1982). The syntheses were performed manually starting with BOC-trp-resin (3.1 g containing 0.59 mequivalent tryptophan/g). Each amino acid was added with the following steps: (a) Resin washings with 3×30 ml CH_2Cl_2 . (b) Removal of the protecting group in 30 ml 33% trifluoroacetic acid (Merck, Darmstadt, FDR), 5% ethanedithiol (Aldrich-Chemie, Steinheim, FDR), in CH_2Cl_2 , containing 1% (wt/vol) indole (Sigma Chemical Co.). Two incubations were used, the first for 5 min and the second for 25 min. (c) Washings with 6×30 ml CH_2Cl_2 . (d) Incubation for 5 min with two 30-ml aliquots of 5% (vol/vol) di-isopropylethylamine in dichloromethane. (e) Washings with 6×30 ml CH_2Cl_2 . (f) Addition of the BOC-amino acid (see below). (g) Washings with 6×30 ml CH_2Cl_2 .

After step (g) the extent of coupling was determined using a quantitative ninhydrin assay (Sarin et al., 1981).

The conditions used for coupling the amino acids varied. Initially a 2.5-fold excess of the BOC-amino acid in dimethylformamide (DMF) (BDH Chemicals, Poole, England) was added and stirred with the resin for 10 min. A 2.5-fold excess of dicyclohexylcarbodiimide (Protein Research Foundation, Osaka, Japan) in CH₂Cl₂ was then added and allowed to react at room temperature. For the first five to six cycles a 2-h incubation sufficed to give >99.8% reaction, however, for subsequent residue additions this incubation was extended to 4 h, and a second overnight coupling with a 1.5-fold excess of BOC-amino acid and dicyclohexylcarbodiimide was required. The ¹³C-labeled BOC-amino acids were added in equivalent molar amounts to the trp on the original resin, and allowed to react in the presence of dicyclohexylcarbodiimide overnight. This step was followed by a 4-h incubation with a 2.5-fold excess of the corresponding unlabeled BOC-amino acid.

The completed peptide was removed from the resin by reaction with ethanolamine, and formylated as described by Prasad et al., 1982. The crude gramicidin A was purified by filtration in methanol on a 100 cm × 3.2 cm diam column of Sephadex LH-20 (Pharmacia, Uppsala, Sweden). Fractions collected from this column were analyzed by high performance liquid chromatography on an analytical (8 mm internal diam × 10 cm), radial-compression column using an isocratic 83:17 (vol/vol) methanol/water solvent. Fractions containing predominantly material with the same HPLC retention time as the major peak from commercial gramicidin A

(Koch-Light, Colnbrook, England), were pooled. Further purification was achieved by preparative thin layer chromatography on 2-mm-thick silica gel plates (Merck Kieselgel 60 F-254). Bands visualized under short wavelength ultraviolet light were extracted from the resin using methanol, and analyzed by HPLC. The fractions containing the purest gramicidin were finally purified by repetition of the Sephadex LH-20 chromatography, or by preparative HPLC in 83:17 methanol/water, using the column described above, followed again by Sephadex LH-20 chromatography. The purity of the gramicidin was assessed to be >98% by HPLC, by 200 MHz ¹H-NMR, and by thin layer chromatography (TLC) (silica gel G) with 85:15:3 CHCl₃/CH₂OH/acetone, as the ascending solvent. TLC spots were detected under UV radiation, or by charring with 25% H₂SO₄.

Sample Preparation

Solutions of gramicidin A and DMPC (1:15 mol ratio) in $C_6H_6/$ CH₃CH₂OH (95:5) were pipetted onto the surface of typically, 50 microscope coverslips, each 15-mm long and a width chosen such that the stacked cross-section was circular with a diameter of <10 mm. The solvent was removed under vacuum overnight. The total weight of lipid plus gramicidin A was ~50 mg. Using a microliter syringe 2–5 μ l of glass distilled water was added to the surface of each coverslip. The coverslips were stacked and the assembly sealed inside a 10-mm diam \times 25-mm long NMR tube using a fine gas flame. Particular care was taken not to damage the dispersion. The sample was equilibrated for 2–3 h at 330 K and then cooled to the temperature at which the measurement was to be made. Longer periods of equilibration failed to improve the alignment of the oriented dispersions.

NMR Spectroscopy

Spectra were acquired using a model CXP300 NMR spectrometer (Bruker Instruments, Inc., Billerica, MA). The NMR tube was mounted horizontally inside a four-turn, close-fitting, 18-gauge, silver-plated copper wire coil tuned to 75.46 MHz for ¹³C and 300.066 MHz for ¹H. Spectra were obtained using the Pines et al. (1973) 13C, proton-enhanced, cross-polarization sequence. Typical operating conditions were: Hartmann-Hahn 90° pulse, 8 µs; repetition delay, 2 s; contact time, 1.5 ms; acquisition time 8.5 ms; sweep width, 60 kHz; 100,000 acquisitions per spectrum, obtained in lots of 10,000. The sample tube was mounted to permit rotation through an angular range of 180° about an axis perpendicular to the Bo field. A goniometer built in this laboratory permitted the angle to be adjusted with the probe and sample in situ. The angle was calibrated from the 'H resonance "magic angle" effect. The reproducibility of the angle setting was ±1°. The water content of the sample was determined from the integrated ¹H-NMR spectrum measured at the magic angle as shown in Fig. 1. The water content was checked before and after each period of acquisition. Owing to the slight (<1 K) heating effects of the radiofrequency (rf) field, a slow migration of water occurs from beneath the rf coil to the ends of the sample tube. If the water content was found to have changed significantly, the sample was reequilibrated. The temperature rise due to rf heating was measured from the apparent depression of the phase transition temperature of a DMPC dispersion.

Spectral Simulations

Spectral simulations were performed on a model HP1000F computer (Hewlett-Packard Co., Palo Alto, CA) using a frequency domain representation of the tensor distribution.

RESULTS

NMR of Dry Powders

Gramicidin A has the sequence HCO-L.Val¹-Gly²-L.Ala³-D.Leu⁴-L.Ala⁵-D.Val⁶-L.Val⁷-D.Val⁸-L.Trp⁹-

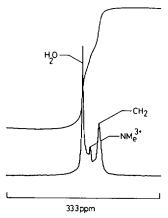


FIGURE 1 300 MHz 1H-NMR spectrum obtained from an aligned sample of gramicidin A and DMPC with the normal to the plane of the multilayers at the "magic angle" (54° 47') to the Bo field. The resulting collapse of the dipolar interactions produces a separation of the H2O, NMe1+ and aliphatic resonances of the sample. Spectrum simulation and comparison of the peak integrals provides a measure of the water content of the dispersion (typically 19-21 mol H₂O/mol DMPC). Acquisi-

tion parameters: sweep width, 10 kHz; 90° pulse duration, 7 μ s; repetition delay, 10 s; 100 acquisitions; temperature, 300 K.

D.Leu¹⁰-L.Trp¹¹-D.Leu¹²-L.Trp¹³-D.Leu¹⁴-L.Trp¹⁵-N HCH₂CH₂OH (Sarges and Witkop, 1965). Dry powders of the gramicidin A analogues with ¹³C—O at Gly², Ala³, or Val⁷, gave the spectra shown in Fig. 2. From these spectra we derived an average value for the principal shielding axes shown in the traceless form of $\sigma_{11} = -74 \pm 3$ ppm, $\sigma_{22} = -7 \pm 3$ ppm, and $\sigma_{33} = +82 \pm 3$ ppm. Within experimental precision the three analogues gave the same shielding tensor. The principal shielding values are close to those obtained from the dipeptide glycyl-glycine · HCl ·

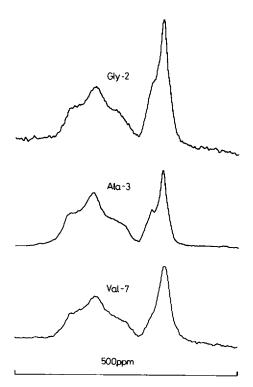


FIGURE 2 75 MHz 13 C cross-polarization NMR spectra of dry powders of 13 C-labeled analogues of gramicidin A with 13 C—O groups at the residues of Gly², Ala³, and Val². Acquisition parameters: sweep width, 62.5 kHz; 90° spin-locking pulse duration, 4 μ s; Hartmann-Hahn contact time, 1 ms; repetition delay, 1 s; broadening, 50 Hz; 10,000 acquisitions; temperature, 273 K.

 H_2O (Stark et al., 1983). Magic angle spinning of dry powdered Ala³-labeled gramicidin A revealed that $\sim 90\%$ of the total spectrum intensity in the low field region of Fig. 2 arose from the carbonyl resonances of the polypeptide backbone.

NMR of Hydrated Dispersions with d₅₄-DMPC

Incorporation of the Val⁷ gramicidin A analogue (1:15) into multilamellar dispersions of chain perdeuterated DMPC (d₅₄-DMPC) yields the spectra shown in Fig. 3 (Smith and Cornell, 1986). The ds4-DMPC analogue of DMPC minimizes the contribution to the cross-polarization ¹³C spectrum of the lipid. Despite the major reduction in the CH₂ peak in Fig. 3 the d₅₄-DMPC carbonyl resonance is only halved in intensity. This indicates that at least half the cross-polarization of the lipid carbonyl groups is derived from the glycerol protons. At temperatures well below the L_{g} - P_{g} phase boundary the carbonyl resonance from the lipid-gramicidin A dispersion indicates a rigid structure. When the temperature is raised to 5° below the d₅₄-DMPC main transition temperature (~290 K at this hydration), the carbonyl resonance narrows, indicating the onset of motion by the gramicidin A. As seen from the d₅₄-DMPC methylene resonance, the lipid has yet to undergo the transition to a fluid state. By 289 K the gramicidin A carbonyl resonance has reached a limiting width while the d₅₄-DMPC methylene resonance reflects a mixed phase. At 297 K the d₅₄-DMPC has fully undergone the main phase transition and peaks due to the d₄-DMPC carbonyls are superimposed on the carbonyl peak from the gramicidin A. Similar results were obtained for the Gly²- and Ala³-labeled compounds.

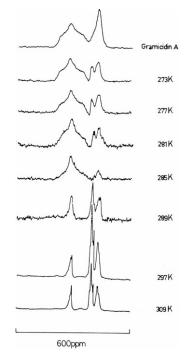


FIGURE 3 75 MHz 13C crosspolarization NMR spectra of the Val⁷-labeled analogue of gramicidin A in a hydrated disperson of perdeuterated d₅₄-DMPC at a mole ratio of 1:15. A temperature independent spectral component thought to be undispersed gramicidin A and amounting to ~20% of the signal intensity was subtracted. Acquisition parameters: sweep width, 62.5 kHz; 90° spin locking pulse duration, 5 μ s; Hartmann-Hahn contact time, I ms below and 2 ms above the main transition temperature; repetition delay, 2 s. The temperature at which each spectrum was acquired is shown adjacent to each curve.

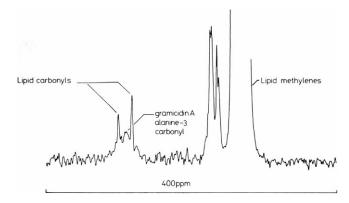


FIGURE 4 75 MHz 13 C cross-polarization NMR spectrum of the Ala³-labeled analogue of gramicidin A dispersed with hydrated multilayers of DMPC (1:15 mol ratio) and aligned on a stack of 50 microscope coverslips. The normal to the plane of the multilayers was directed along the Bo field (0° orientation). Acquisition parameters: sweep width, 62.5 kHz; 90° spin locking duration, 9 μ s; Hartmann-Hahn contact time, 2 ms; repetition delay, 2 s; line broadening, 50 Hz; 23,000 acquisitions; temperature, 308 K.

NMR of Aligned Dispersions with dsa-DMPC and DMPC

High Temperature Phase. The ¹³C spectrum of the Ala³ analogue aligned in DMPC at 307 K is shown in Fig. 4. The sample is oriented with the slide normal parallel to the B_o field (0° orientation). The unresolved carbonyl multiplet seen at high temperatures in Fig. 3 is now separated into two narrow lipid peaks and a broader gramicidin A resonance. The dependence on orientation of these three signals is shown in Fig. 5. The gramicidin A

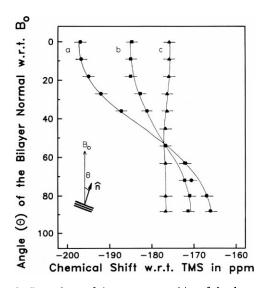


FIGURE 5 Dependence of the resonance position of the three carbonyl peaks seen in Fig. 4, on the orientation angle θ between the normal to the plane of the coverslips and the B_o field. The CSA of (a) the DMPC sn-1 chain ester carbonyl (-30 ± 1 ppm), (b) the Ala³-labeled analogue of gramicidin A (-16 ± 1 ppm), and (c) the DMPC sn-2 chain ester carbonyl (+2 ± 1 ppm).

carbonyl chemical shift anisotropy (CSA) may now be measured as -16 ppm. Fig. 6 shows the carbonyl regions at orientations with the normal to the slide at 0°, the "magic angle" and 90° relative to the B_o field. Lorentzian simulations of the peaks and the de-convoluted contribution of the gramicidin A are shown adjacent to each spectrum. The half height peak width of the gramicidin A 13 C=O resonance of 1,900 ± 150 Hz is constant from 0° to the "magic angle" but falls to 900 ± 150 Hz at 90°. Similar results were obtained for the Val⁷ analogue. Aligned dispersions containing the even numbered Gly² analogue have a ${}^{13}C = O CSA$ of -9 ppm. Examples of the spectrum at 0° orientation for the single labeled analogues are shown in Fig. 7. The lipid carbonyl resonances from the Val⁷ analogue are smaller than that from the other dispersions owing to the use of ds4-DMPC. The use of ds4-DMPC did not prevent the sn-2 lipid carbonyl peak obscuring the signal from the Gly² analogue, making it difficult to obtain an accurate estimate of the CSA. In Fig. 8 we show that the ¹³C=O peak of the Gly² analogue may be resolved by using ether-linked lipids. The ester carbonyl resonance from the lipid is lost and an additional peak is evident in the glycerocholine region of the spectrum due to the ether methylene carbon. Use of ether-linked lipids permits a comparison of the spectrum of the single and double ¹³C-labeled Gly² analogue. As seen in Fig. 9 the dipolar interaction of the alpha and carbonyl carbons produce a doublet with a splitting of 1,900 ± 150 Hz.

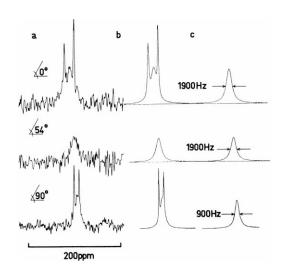


FIGURE 6 (a) Carbonyl region of the 75 MHz, 13 C cross-polarization NMR spectrum of Ala³-labeled gramicidin A analogue in an aligned, hydrated disperson of DMPC. The normal to the plane of the multilayers was oriented at 0°, the "magic angle" (54° 47′), and 90° to the B₀ field. Acquisition parameters: sweep width, 62.5 kHz; 90° spin locking pulse duration, 8 μ s; Hartmann-Hahn contact time, 2 ms; repetition delay, 2 s; line broadening, 50 Hz; 60,000 (0°), 100,000 ("magic angle"), and 70,000 (90°) acquisitions; temperature, 308 K. (b) Lorentzian simulations from which it was possible to deconvolute the contribution of the gramicidin A 13 C—O groups. (c) The deconvoluted spectra from the gramicidin A 13 C—O groups. The line widths at 0°, "magic angle" and 90° were 1,900 \pm 150, 1,900 \pm 150, and 900 \pm 150 Hz, respectively.

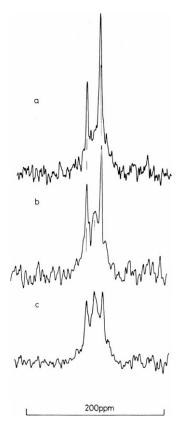


FIGURE 7 Comparison of the 0° orientation of gramicidin A analogues dispersed in DMPC (1:15) with ¹³C—O labels at (a) Gly², (b) Ala³, and (c) Val³. Whereas the sn-1 and sn-2 carbonyl peaks from the lipid are well resolved from Val³ and Ala³ carbonyl resonances, the Gly² carbonyl peak is obscured. The acquisition parameters were similar to those in Figs. 4 and 6 for the 0° orientation; temperature, 308 K.

Low Temperature Phase

Cooling the aligned dispersion to 274 K causes no change in the shape or position of the gramicidin A carbonyl spectra obtained from samples at the 0° orientation. An example of this is seen in a comparison of the high and low temperature 0° orientations for the Gly² analogue, shown in Figs. 8 and 10. From Fig. 10 it is evident that the behavior of the spectrum at other orientations is very different when the sample is cooled such that the lipid is in the L_B phase. The peak broadens substantially as the angle of the slide normal approaches 90° to the B_0 field. At 308 K with the lipid in the L_{α} phase the same change in angle primarily causes a shift in position and an approximately twofold reduction in peak width. Simulations of these effects are given in a following section.

DISCUSSION

The configuration of gramicidin A has been shown to depend upon its environment. In dioxane solution two forms exist. One of these, termed species 3, is a $\downarrow \uparrow \pi \pi^{5.6}_{LD}$ antiparallel left-handed, double-stranded helical dimer. The other, species 4 is a $\uparrow \uparrow \pi \pi^{5.6}_{LD}$ parallel right-handed, double-stranded helical dimer (Veatch et al., 1974; Sychev et al., 1980; Arseniev et al., 1984). In sodium dodecyl sulfate (SDS) micelles, gramicidin A forms a single-stranded $\pi \pi^{6.3}_{LD}$ head-to-head, right-handed, helical dimer (Arseniev et al., 1985b). In dispersions of lysophosphatidyl-choline, it has been shown that a single-stranded

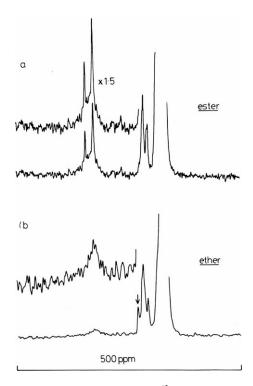


FIGURE 8 Comparison of the 75 MHz, 13 C cross-polarization NMR spectra of the Gly²-labeled gramicidin A analogue dispersed in either (a) hydrated dimyristoyl (ester-linked chains) or (b) dimyristyl (ether-linked chains) phosphatidylcholine, at a mole ratio of 1:15. The dispersions were aligned with the normal to the multilayers along the B_0 field. The arrow in spectrum b denotes the peak arising from the methylene carbon adjacent to the ether oxygen. The inset in a has a larger vertical gain by a factor of 1.5. Acquisition parameters: sweep width 62.5 kHz; 90° spin locking duration, 7 μ s; Hartmann-Hahn contact time, 1.5 ms; repetition delay, 2 s; line broadening, 50 Hz; 30,000 acquisitions; temperature, 308 K.

 $\vec{\pi} \stackrel{\leftarrow}{\pi}^{6.3}_{LD}$ head-to-head, left-handed, helical dimer is formed (Urry et al., 1982, 1983a, b). When complexed with cesium ions and dissolved in a methanol-chloroform solution yet another structure is found (Arseniev et al., 1985a). This is a double-stranded $\downarrow \uparrow \pi \pi^{j,2}_{LD}$ antiparallel righthanded double helical dimer. Many techniques, including, infra-red, fluorescence and circular dichroism spectroscopy (Veatch et al., 1974; Veatch and Blout, 1974; Sychev et al., 1980; Wallace et al., 1981; Nabedryk et al., 1982; Urry et al., 1983a, b; Cavatorta et al., 1982; Sychev and Ivanov, 1984; Naik and Krimm, 1984; Wallace, 1984, 1986; Urry, 1984), solution NMR (Fossel et al., 1974; Urry et al., 1979, 1982; Weinstein et al., 1979, 1980, 1985; Arseniev et al., 1984, 1985a, b; Hawkes et al., 1984), and x-ray (Koeppe et al., 1978; Kimball and Wallace, 1984; Wallace, 1986) and neutron diffraction (Koeppe and Schoenborn, 1984) have been used to probe the conformation of gramicidin A. Yet none of these approaches has provided a unique conformation for gramicidin A in phospholipid bilayers. The structural detail available through the use of high resolution NMR techniques is overshadowed by the need to use an isotropic solvent system.

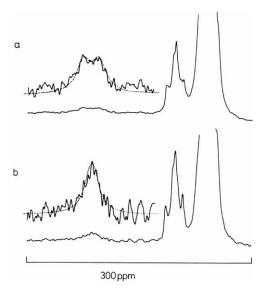


FIGURE 9 Comparison of the 75 MHz, 13 C cross-polarization NMR spectra of an aligned, hydrated dimyristylphosphatidylcholine (ether linked) dispersion of the Gly²-labeled gramicidin A analogue, (1:15) with the spectrum from a similar dispersion containing the double-labeled Gly² (13 C \longrightarrow 0, 13 C_a-carbon) analogue. The normal to the multilayers was oriented along the B_o field. The spectra are superimposed on Lorentzian functions which, for the single-labeled compound is a single Lorentzian with a full width half-height of 1,900 Hz, and for the double-labeled compound is two Lorentzians of the same width but a separation of 1,900 Hz. The insets in the carbonyl region of the spectrum have an eightfold increase in the vertical scale. Acquisition parameters: sweep width, 62.5 kHz; 90° spin locking pulse, 7 μ s; Hartmann-Hahn contact time, 1.5 ms; repetition delay, 2 s; line broadening, 50 Hz; 150,000 (double label) and 50,000 (single label) acquisitions; temperature, 308 K.

Techniques such as infrared or circular dichroism spectroscopy are currently limited to providing a "fingerprint" of the class of structure adopted by the gramicidin A. X-ray and neutron diffraction require single crystals that may possess a different molecular geometry to the dispersion structure. For example, it has been found that the structure of the cesium complex determined by x-ray diffraction is different from that of gramicidin A in dispersions with phospholipid (Wallace, 1986). Furthermore, it has been shown to possess an opposite handedness to gramicidin A—cesium complex dissolved in chloroform-methanol (Arseniev et al., 1985a).

Despite the absence of direct evidence, a combination of techniques has produced a highly constraining set of requirements to which any model for the configuration of gramicidin A in phospholipid bilayers must comply. To date, three types of structure have been proposed. These are the parallel and antiparallel double-stranded helical dimers (Veatch et al., 1974; Arseniev et al., 1984), the end-to-end, single-stranded helical dimers (Urry et al., 1982, 1983), and aggregates of β sheet (Sychev and Ivanov, 1982, 1984).

We have used ¹³C-NMR of gramicidin A in aligned phospholipid bilayers to measure the bond directions of representative carbonyl groups. A starting point for our

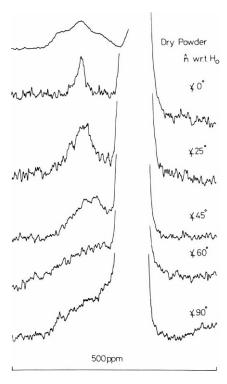


FIGURE 10 75 MHz ¹³C cross-polarization NMR spectra of the Gly² labeled analogue of gramicidin A dispersed (1:15) with aligned DMPC multilayers in the $L_{\rm ff}$ phase at 274 K, and orientations of the multilayer normal at 0°, 25°, 45°, 60°, and 90° to the B_o field. The rigid powder pattern of the Gly² labeled analogue is redrawn from Fig. 2. Acquisition parameters: sweep width, 62.5 kHz; 90° spin locking time, 6 μ s; Hartmann-Hahn contact time, 1 ms; repetition delay, 1 s; line broadening, 50 Hz; typically 100,000–180,000 acquisitions.

analysis is to note that when the aligned dispersions are cooled and the gross molecular motion eliminated, the carbonyl spectra at the 0° orientation are unaltered. This indicates that the polypeptide has a symmetry axis directed along the bilayer normal. This result also demonstrates the absence of substantial fluctuations in the alignment of gramicidin A relative to the plane of the bilayer or in its internal geometry. This does not exclude small amplitude bond vibrations that may be independent of the compressibility of the surrounding lipid. A recent deuterium NMR study by Datema et al., 1986, has observed the onset of axial rotation by gramicidin A on heating through the lipid phase transition.

Treating gramicidin A as either a rigid sheet or cylinder we may test the predicted CSA of the current models against the experimental NMR data. Figs. 11-13 show the direction of the peptide carbonyl bonds relative to the symmetry axis appropriate to each model. The helix axis is taken as the symmetry axis in both the single- and double-stranded helix models. For the β sheet aggregates, we have taken the long axis of the "J-hairpin" (Sychev and Ivanov, 1982) as the symmetry axis. As seen in Fig. 11 there is a common orientation for all of the carbonyl bonds in the β sheet aggregate. In the helical structures, the carbonyl

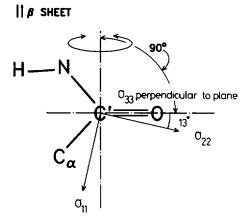


FIGURE 11 Common orientation of all amide carbonyl bonds relative to the long axis of the "J" hairpin structure, proposed (Sychev and Ivanov, 1982) for gramicidin A.

bond direction depends upon whether the amino acid is an odd- or even-numbered residue. Fig. 12 shows the two sites found in the parallel and antiparallel double-stranded $\pi\pi_{LD}$ helices, and Fig. 13 the two sites for the single-stranded $\pi\pi_{1D}$ helices. The predicted CSA for each of these structures assuming rapid reorientational averaging about the symmetry axis is given in Table I. These predictions are contrasted with the measured values from the three analogues in dispersions at 308 K with the lipid in the L_a phase. The double-stranded helices and the β sheet configurations fail to predict the experimental CSA. The data are best matched by assigning the Gly² label to type B site and the Ala³ and Val⁷ labels to the type A site of a single-stranded left-handed $\vec{\pi}$ $\vec{\pi}^{6.3}$ _{LD} helix. Agreement with the single-stranded right-handed $\vec{\pi} \stackrel{\leftarrow}{\pi}^{6.3}_{LD}$ helix requires the reversal of the assignment of the type A and B sites. The difference in direction of the carbonyl bonds between these two cases is only 15°. Given the uncertainty in the detailed

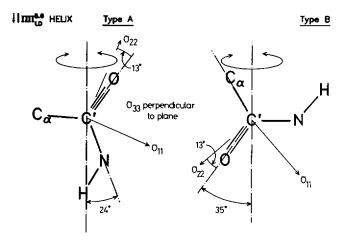


FIGURE 12 Orientation of type A and type B carbonyl sites relative to the helix axis for the double-stranded $\pi\pi^{56}_{LD}$ helix proposed (Veatch et al., 1974; Sychev et al., 1980; Arseniev et al., 1984) for gramicidin A. The assignment of the sites to odd- or even-numbered amino acid residues depends on the details of the helix (see text).

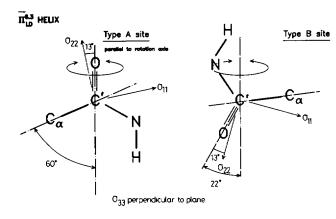


FIGURE 13 Orientation of type A and type B sites relative to the helix axis for the single-stranded $\pi\pi^{63}_{LD}$ helix proposed (Urry et al., 1982, 1983) for gramicidin A. The assignment of the sites to the odd or even amino acid residues depends on the details of the helix (see text).

conformation of these structures and the uncertainty in the orientation of the 13 C shielding tensor (typically $\pm 5^{\circ}$) we are reluctant to exclude the latter geometry. Further support for this general interpretation was obtained from the double-labeled Gly² analogue of gramicidin A. The residual dipolar coupling for the ¹³C—¹³C interaction between the alpha and carbonyl carbons is shown in Table II for a selection of geometries. Again, the single-stranded, left-handed $\vec{\pi} \stackrel{\leftarrow}{\pi_{1D}}$ helix is consistent with the measured data. The disorder of the symmetry axis is also seen to be small with a simple axial reorientation accounting for the reduced dipolar coupling. In total there are four singlestranded $\vec{\pi} = \vec{\pi}^{6.3}_{LD}$ helices which may need to be considered. In the left-handed helix described by Urry et al., 1980, the type A sites are associated with the odd-numbered residues and the type B sites with the even-numbered residues. In the right-handed helix described by Arseniev et al., 1985, the assignment is reversed. Two more single-stranded helices may be generated by reversing the helix handedness while maintaining the assignment of the type A and B sites as before. However, from wire models of these structures, they may be energetically unfavorable due to the steric

TABLE I
THE PREDICTED RESIDUAL CSA FOR THE ¹³C — O
RESONANCE IN A LABELED ANALOGUE
OF GRAMICIDIN A WHICH IS RAPIDLY
REORIENTING ABOUT THE SYMMETRY
AXIS APPROPRIATE TO EACH STRUCTURE

Structure	CSA -102 ppm	
β sheet		
ππ ^{5.6} LD	-26 ppm (Type A)	-65 ppm (Type B)
$\pi\pi^{6.3}_{LD}$	-17 ppm (Type A)	-13 ppm (Type B)
Experimental result	-16 ± 1 ppm (Ala³, Val²)	-9 ± 1 ppm (Gly²)

The orientation and magnitude of the $^{13}C = O$ tensor was taken from Stark et al., 1983, as -74 ppm, -7 ppm, and 82 ppm, for σ_{11} , σ_{22} , and σ_{33} , respectively, with σ_{33} perpendicular to the plane of the O-C = O group and σ_{22} tilted by 13° from the C = O bond direction.

TABLE II

THE PREDICTED RESIDUAL ¹³C-¹³C DIPOLAR INTERACTION BETWEEN THE ¹³C = O AND α-CARBONS IN THE Gly²-LABELED ANALOGUE OF GRAMICIDIN A, ASSUMING RAPID REORIENTATION ABOUT THE GRAMICIDIN SYMMETRY AXIS

Structure	Dipolar splitting at 0° orientation		
β sheet	2,375 Hz		
$\pi\pi^{5.6}_{LD}$	1,830 Hz (Type A)	2,860 Hz (Type B)	
ππ ^{6.3} LD	475 Hz (Type A)	1,857 Hz (Type B)	
Experimental			
result	$1,900 \pm 150 \mathrm{Hz}$		

The dipolar coupling constant was calculated from $\gamma \ell r^{-3}$ as 2,087 Hz using a C-C bond distance of 0.1540 nm. The 90° discontinuities in the ¹³C powder pattern of the double labeled glycine from which the labeled gramicidin A was synthesized had a separation of 1,900 Hz.

interference of some of the amino acid side chains with the polypeptide backbone. The angular dependence seen for the dispersions in the fluid phase excludes macroscopic aggregation of the gramicidin A. However, aggregates on the molecular scale could reorient sufficiently rapidly to give the observed axially symmetric spectra.

A result that requires further explanation is the width of the peaks seen in Figs. 8 and 10 for the 0° orientation. As this width does not alter on freezing of the lipid, it is unlikely to be dominated by either relaxation effects or by a slow dynamic disorder. For the same reason we discount dipolar coupling to the amide ¹⁴N. Mosaic spread of the sample alignment is unlikely to be the mechanism. An intrinsic check of the bilayer mosiac spread is available in the lipid linewidth, which is five to six times narrower than that from the gramicidin at all angles.

It should be noted that the carbonyl signal from the gramicidin-containing dispersion at the "magic angle" (Fig. 6) is two to three times less intense than at 0°, and could arise from nonaligned material normally not visible. The nonaligned lipid peaks are weak at the contact time of 2 ms used to obtain these spectra.

A measurement of the 13 C T_2 at the 0° orientation of the Ala³ analogue gave 700 \pm 100 μ s. This accounts for less than one-third of the linewidth of the gramicidin A carbonyl peaks and eliminates relaxation as the major cause of the broad carbonyl resonance from the gramicidin.

One possible source of the linebroadening is a distribution of molecular geometry within the gramicidin A. This may include minor changes in the number of residues per turn or even the handedness of the helix. The direction of the carbonyl bonds change by only 22° with a change in the handedness of the helix. The free energy difference between the left- and right-handed configurations is ~2 kcal/mol (Venkatachalam and Urry, 1983). A recent study by Weinstein et al., 1985, has reported a configurational heterogeneity for a series of ¹⁹F-labeled gramicidin A analogues dispersed in DPMC vesicles. Although these authors concluded that the accessibility of these labels was

only consistent with the head-to-head, single-stranded $\vec{\pi} \cdot \vec{\pi}_{LD}$ helix structure for gramicidin A, they obtained two chemical shifts for the C-terminal fluorine labels on the outer leaflet of the DMPC vesicle. The origin of this effect may reflect the same structural heterogeneity seen in the present results.

One test of the proposed model of configurational heterogeneity is to observe the dependence of the lineshape on the slide orientation when the gramicidin A is rigid (Fig. 10). Simulations based on an arbitrary gaussian variation in the ¹³C=O bond direction (SD = 10°) are given in Figs. 14 and 15. The agreement with the experimental results seen in Fig. 10 is encouraging. The model also accounts for the lineshapes at the 0° and 90° orientations when the lipid is in the L_{α} phase (Fig. 5). At the 90° orientation in the L_{β} phase the greatly increased width of the spectrum from the aligned component reduces the signal intensity to a level that is similar to that of the powder signal. From Fig. 15 it is apparent that in the L_{β} phase only ~70% of the gramicidin A is well aligned.

The spectra shown in Fig. 14 are insensitive to the details of the heterogeneity and refinement of this model awaits the results obtained from other labeled analogues.

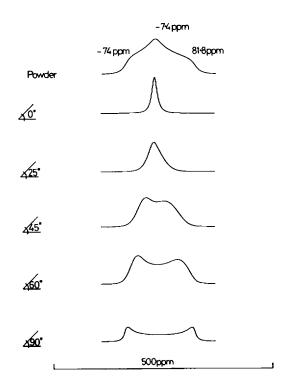


FIGURE 14 Computer generated simulations of the Gly²-labeled gramicidin A dispersion in which the gramicidin A molecules are rigid, although aligned with the helix axis perpendicular to the plane of the multilayers. The orientation and amplitude of the CSA tensor was taken from Stark et al., 1983, with σ_{22} 13° off the C—O bond direction and σ_{33} perpendicular to the plane of the peptide bond. The simulation includes a gaussian (SD = 10°) distribution of the C—O bond direction and a Lorentzian linebroadening equivalent to a T_2 of 700 μ s. The scale and angles have been chosen to allow easy comparison with the data in Fig. 10.

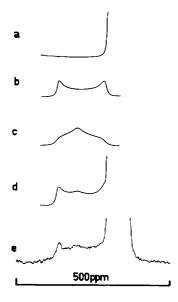


FIGURE 15 Computer simulations of a 75 MHz, ¹³C crosspolarization NMR spectrum of Ala3-labeled gramicidin A analogue in an aligned, hydrated dispersion of DMPC at 274 K. The normal to the plane of the multilayers was oriented at 90° to the Bo field. The simulation is compared with an experimental spectrum obtained using the conditions given in the caption to Fig. 10. Contributions are included from (a) the lipid methylene chains, (b) the aligned Ala3labeled gramicidin A carbonyls, and (c) the nonaligned Ala³labeled gramicidin A carbonyls. (d) Simulation generated from $1.5 \times (a) + 1.0 \times (b) + 0.5 \times (c)$. (e) Experimental spectrum.

The present study of the carbonyl bonds has emphasized the whole molecular motion of gramicidin A. In future reports we will address the orientation of the side chains and the effect of ions on the carbonyl bond directions.

CONCLUSION

The results of this study permit us to make the following observations of a 1:15 gramicidin A per DMPC dispersion: (a) that gramicidin A intercalates into multilayers of DMPC with a geometry in which the carbonyl bond directions are broadly consistent with a $\vec{\pi} = \vec{\pi}^{3}$ helix. The presence of what we interpret as a significant spread in direction about the mean suggests that the helices may possess both left- and right-handedness or a variation in pitch. The double-stranded helices and β sheet structures are excluded. (b) When the DMPC is in the L_{α} phase the gramicidin A reorients about the long axis of the helix at a rate that exceeds the CSA (at 75 MHz) of $\sim 14,000 \,\mathrm{s}^{-1}$. (c) When the DMPC is cooled to the L_{β} phase the gramicidin A becomes rigid although its internal configuration and alignment relative to the bilayer remains unaltered. (d) On heating, motion of the gramicidin A is observed before the main transition temperature of the DMPC.

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